

THE CONSTITUENTS OF SCHIZANDRA CHINENSIS BAILL.

THE STRUCTURE OF A NEW LIGNAN, GOMISIN D

Yukinobu Ikeya and Heihachiro Taguchi\*

Tsumura Laboratory, Izumi, Komae-shi, Tokyo, Japan

Yoichi Iitaka

Faculty of Pharmaceutical Sciences, University of Tokyo

Bunkyo-ku, Tokyo, Japan

(Received in Japan 12 February 1976; received in UK for publication 17 March 1976)

In the previous paper, we reported the structures of three new schizandrin type lignans, gomisin A, B and C, isolated from the fruits of Schizandra chinensis Baill. (Schizandraceae).<sup>1)</sup> The present paper concerns the structure of a new schizandrin type lignan, gomisin D (yield, 0.016 %), isolated from the fruits of the same plant.

Gomisin D (1): colourless prisms (from n-hexane-ether), mp 194°.  $(\alpha)_D^{25} - 58.8^\circ$  ( $c=0.265$ ,  $\text{CHCl}_3$ ),  $\text{C}_{28}\text{H}_{34}\text{O}_{10}$  (m/e,  $\text{M}^+$ , Calcd., 530.2156; Observed, 530.2144), UV  $\lambda_{\text{max}}^{\text{EtOH}}$  (log  $\epsilon$ ): 216(4.57), 256(sh, 3.94), 294(3.68), IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3505, 3450 (OH), 1720 (ester), PMR ( $\delta$  in  $\text{CDCl}_3$ ): 1.02(3H, d,  $J=7$  Hz, H-17), 1.10(3H, d,  $J=7$  Hz, H-24), 1.20(3H, s, H-18), 1.26(3H, s, H-23), 1.72(1H, m, H-8), 1.73(1H, s, OH), 2.0-2.6(2H, ABX octet,  $J_{\text{AB}}=14$  Hz,  $J_{\text{AX}}=8$  Hz,  $J_{\text{BX}}=1$  Hz, H-9), 3.10(1H, s, OH), 3.56, 3.86, 3.92 (each 3H, s, 3 x  $\text{OCH}_3$ ), 3.4-4.0(2H, m, H-22), 5.71(1H, s, H-6), 5.90, 5.99 (each 1H, d,  $J=1$  Hz,  $-\text{OCH}_2\text{O}-$ ), 6.43(1H, s, H-11) and 6.80(1H, s, H-4).

These spectral data suggested that gomisin D is a schizandrin type lignan, having a methylenedioxy group ( $-\text{OCH}_2\text{O}-$ ), a methyleneoxy group ( $-\text{CH}_2\text{O}-$ ) and three methoxyl groups attached to the aromatic ring and also has a ester linkage and two tertiary hydroxy groups.<sup>1-3)</sup>

Hydrolysis of 1 with 3 % ethanolic potassium hydroxide followed by methylation with diazomethane afforded a crystalline substance (2),  $\text{C}_{29}\text{H}_{38}\text{O}_{11}$  (m/e,

$M^+ - H_2O$ , Calcd. 544.2854; Observed, 544.2809), mp 127-128°,  $(\alpha)_D^{22} +39.1^\circ$  (c=0.307,  $CHCl_3$ ). UV  $\lambda_{max}^{EtOH}$  nm (log  $\epsilon$ ): 217(4.68), 255(sh, 3.97), 278-280(sh, 3.53). IR  $\nu_{max}^{KBr}$   $cm^{-1}$ : 3430, 1730. 2 was oxidized with sodium periodate to show the presence of a 1,2-diol. The presence of a partial structural system (A) in 2 was confirmed by the double resonance experiment of PMR (in  $C_6D_6$ ) and the mass spectral analysis (chart 2).

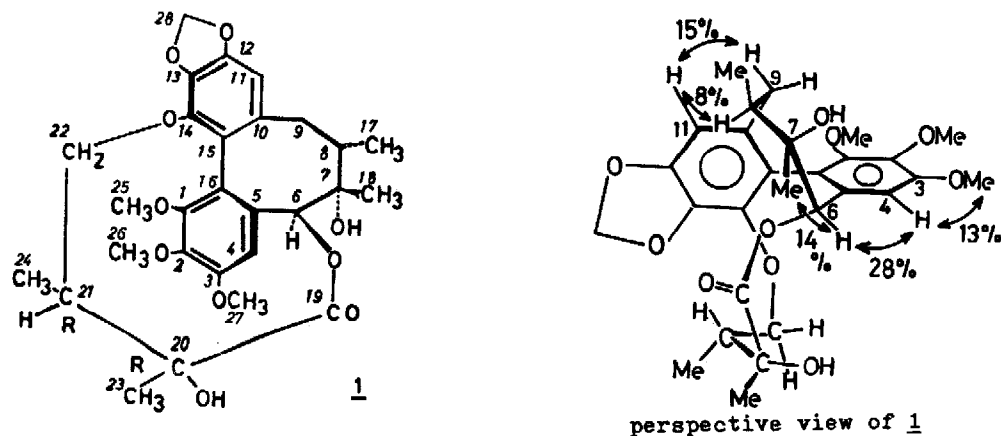


Chart 1

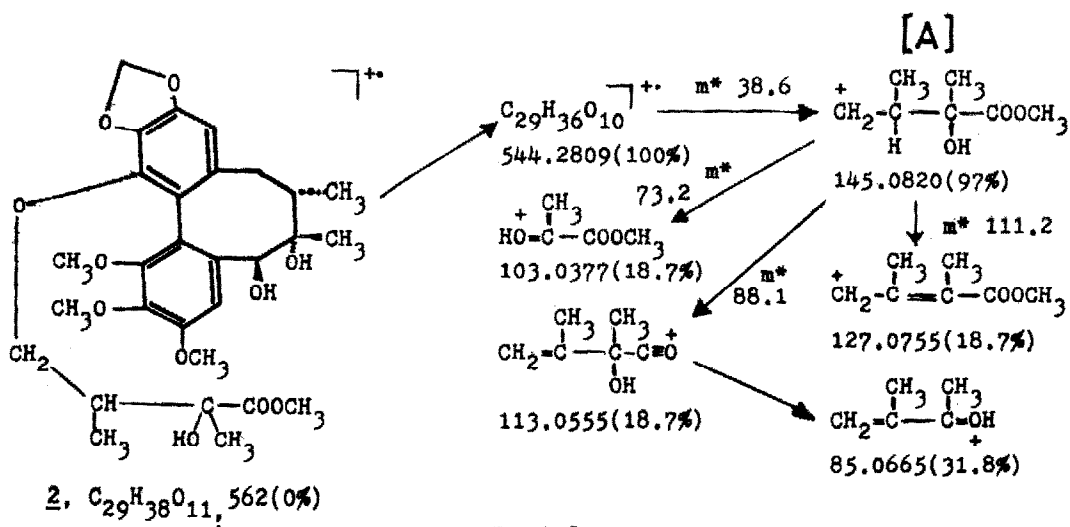


Chart 2

These data indicate that the carboxyl terminal of (A) is linked as ester in 1. The conformational structure of 1 was thus elucidated by the measurements of NOE in 1 (in  $\text{CDCl}_3$ )<sup>1,2)</sup> As shown in chart 1, irradiation of a methoxyl signal ( $\delta$  3.86) and a proton ( $\delta$  5.71) caused 13 % and 28 % increase in the integrated intensity of the lower aromatic proton, respectively. On the other hand, irradiation of a tertiary methyl signal ( $\delta$  1.20) caused 14 % increase in the integrated intensity of a proton ( $\delta$  5.71), while no enhancement of the signal intensity of the both aromatic protons was observed. A higher aromatic proton ( $\delta$  6.43) was effected by irradiation of a higher field proton of methylene signal ( $\delta$  2.00, d,d,  $J=14/1$  Hz) and a methine proton ( $\delta$  1.72, H-8), while no effect was shown by irradiation of each methoxyl signal and the doublet methyl signals. From the above results and the Dreiding model examination, the structure of gomisin D was suggested to be 1 excluding the configurations at  $\text{C}_{20}$  and  $\text{C}_{21}$ .

Finally, the structure including the absolute configuration of gomisin D was elucidated by X-ray crystallographic analysis of the 4,11-dibromo-derivative (3). Treatment of 1 with bromine in carbon tetrachloride saturated with water afforded 3 as colourless plates (from ether-petroleum ether). The crystals are orthorhombic, space group  $P 2_1 2_1 2_1$  with  $a=14.756(7)$ ,  $b=20.172(10)$ ,  $c=9.854(5)$  Å,  $U=2933.13$  Å<sup>3</sup>,  $D_m=1.54$  g.cm<sup>-3</sup>,  $D_x=1.513$  g.cm<sup>-3</sup>,  $Z=4$ . The density was measured by the flotation method using an aqueous potassium iodide solution. The intensity data were measured by a Philips four-circle diffractometer using graphite monochromated Cu K $\alpha$  radiation. The space group was determined by the Weissenberg photographs. Of the total 3119 independent reflections within  $156^\circ \leq 2\theta$ , 3036 reflections were used for the structure determination. Intensities of 1074 Friedel pairs were also measured on the diffractometer. The structure was solved by the heavy atom method using three dimensional Patterson map and Fourier synthesis. Refinement was carried out by the block-diagonal least-squares method using HBL5 programme<sup>4)</sup>. Final R value was 0.089 without hydrogen atoms. The absolute structure was established by using the anomalous dispersion effect of the bromine atom for Cu K $\alpha$  radiation.

Of the 127 Friedel pairs for which  $||F_{\text{obs}}(\text{hkl})| - |F_{\text{obs}}(\text{h}\bar{k}l)||$  were estimated to be greater than  $3\sigma(F_{\text{obs}}(\text{hkl}))$ , 119 pairs showed the consistent  $F_{\text{obs}}$  and  $F_{\text{cal}}$  relations of the Friedel pairs and the absolute configuration was determined as shown in chart 3. The molecular structure of **3** drawn by the ORTEP programme is shown in chart 3. The bond lengths and the angles not including hydrogen atoms lie in the normal ranges. The configuration of each carbon atom is shown as 6S, 7S, 8S, 20R and 21R, respectively, and the biphenyl part is shown to be S-configuration<sup>5)</sup> with dihedral angle of 73°. Furthermore, the conformation of cyclooctadiene ring is shown to be twist-boat-chair form.

These data are in good agreement with the above physical data of gomisin D. Therefore, the structure of gomisin D is represented by **1**.

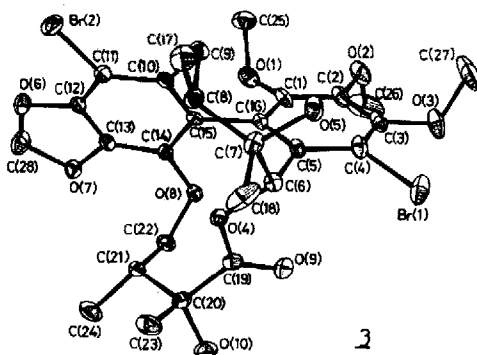


Chart 3

**ACKNOWLEDGEMENT:** The authors express their gratitude to Prof. S. Shibata, Dr. U. Sankawa and Dr. T. Akiyama, Faculty of Pharmaceutical Sciences, University of Tokyo, for their valuable discussions during this work. Thanks are also due to Prof. S. Fukushima, Shizuoka College of Pharmacy, and Prof. T. Yamada, Tokyo College of Pharmacy, for mass spectral measurements.

#### REFERENCES

- 1) H. Taguchi and Y. Ikeya, Chem. Pharm. Bull. (Tokyo), **23**, 3296 (1975).
- 2) Y.-P. Chen, R. Liu, H.-Y. Hsu, S. Yamamura, Y. Shizuri and Y. Hirata, Tetrahedron Letters, **1973**, 4257.
- 3) N. K. Kochetkov, A. Khorlin, O. S. Chizov and V. I. Sheichenko, Tetrahedron Letters, **1961**, 730 and references cited therein.
- 4) Y. Okaya and T. Ashida (1967). HBLV IV. The Universal Crystallographic Computing System (I), p. 65. The Crystallographic Society of Japan.
- 5) R. S. Cahn, C. K. Ingold and V. Prelog, Experientia, **12**, 81 (1956).